Tetranortriterpenoids from Ruagea glabra

Baldwin S. Mootoo* and Russel Ramsewak

Department of Chemistry, University of the West Indies, St. Augustine, Trinidad and Tobago

Ayub Khan

Department of Plant Sciences, University of the West Indies, St. Augustine, Trinidad and Tobago

Winston F. Tinto

Institute of Marine Affairs, Hilltop Lane, Chaguaramas, P.O. Box 3160, Carenage Post Office, Trinidad and Tobago

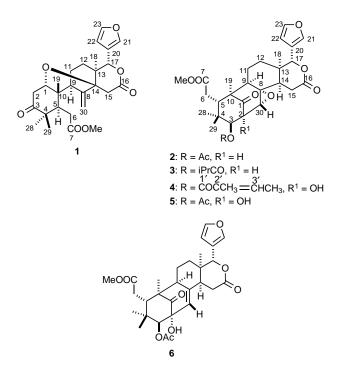
William F. Reynolds, Stewart McLean, and Margaret Yu

Department of Chemistry, University of Toronto, Toronto, Ontario M5S 1A1 Canada

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Six tetranortriterpenoids including four new compounds, ruageanins A (**3**), B (**4**), C (**5**), and D (**6**), were isolated from fruits of *Ruagea glabra* (Meliaceae). Structures of the new compounds were established by analysis of the high-field NMR data. Methyl angolensate (**1**), xylocarpin (**2**), and ruageanins A (**3**) and B (**4**) showed significant antifeedant activity to the final instar larvae of *Spodoptera frugiperda*.

In our continuing studies¹ on the Meliaceae of tropical and semitropical America, we have investigated Ruagea glabra Triana and Planchon, a member of the Meliaceae which is closely related to Guarea species. The plant material was collected at an altitude of 2060 m in Cartago Costa Rica. We obtained no limonoids from the ground heartwood. However, six compounds of this type were isolated and characterized from the seeds, viz. the known compounds methyl angolensate (1)^{2,3} and xylocarpin $(2)^4$ and the new bicyclononalides ruageanins A (3), B (4), C (5), and D (6). We report here the identification of these compounds and the structure elucidation of 3-6. The latter three compounds are members of the mexicanolide group of limonoids possessing the relatively rare –OH or O-acyl C-2 substituent.



* To whom correspondence should be addressed. Tel./Fax: (809) 645-7132. E-mail: mootoo@chem.uwi.tt. [®] Abstract published in *Advance ACS Abstracts*, April 15, 1996.

Methyl angolensate (1) and xylocarpin (2) were identified by a comparison of their mp, IR, and ¹H-NMR data with the reported values,²⁻⁴ as well as by a detailed analysis of the ¹H and ¹³C spectral data (Tables 1 and 2). In both cases the bond connectivities were clearly established.

The ¹H and ¹³C NMR data for the new compounds ruageanin A–D also allowed unambiguous assignment of the signals (Tables 1 and 2), and several ${}^{2,3}J_{CH}$ bond connectivities could be established for most of the carbon atoms.

Ruageanin A (3), $C_{31}H_{40}O_9$ (HRMS), showed absorption maxima in the IR spectrum for carbonyl (1730 cm⁻¹) and furan (875 cm⁻¹). The ¹H and ¹³C NMR data were almost identical to those of xylocarpin,⁴ including the characteristic proton on the epoxide ring (δ_H 3.31, d, 2.0 Hz). The substituent at C-3 now present as an isobutyryl moiety showed δ_H 1.26 (3H, d, 7.0 Hz), δ_H 1.28 (3H, d, 7.0 Hz), and δ_H 2.75 (1H, m) with the corresponding δ_C signals at 18.94, 19.45, 34.17, and 176.02. The detailed bond connectivities and assignment of all the NMR signals (¹H and ¹³C) were obtained from the COSY, HMBC, and HETCOR data.

Ruageanin B (4) was obtained as a crystalline solid by repeated recrystallization from a mixture with humulin B.⁷ The molecular formula, $C_{32}H_{40}O_{10}$, was obtained from HRMS data. Ruageanin C (5) was also crystalline and showed a molecular formula C₂₉H₃₆O₁₀. The ¹H NMR spectra of these two compounds are virtually identical except that ruageanin B bears an *O*-tigloyl group at C-3 which is replaced in compound **5** by an acetate. H-3 and H-30 are characteristic singlets in these molecules (for compound 5 at δ 5.09 and 3.52, respectively), and all the ¹H and ¹³C assignments could be made from the NMR data. These compounds are closely related to humilinolides A-D recently reported from Swietenia humilis⁸ and are 6-deoxy analogues of humilinolide A. The structure and stereochemistry of the latter was confirmed by single-crystal X-ray diffraction studies.8

Ruageanin D (**6**), the least abundant of the compounds isolated, was obtained as an amorphous solid. The molecular formula, $C_{29}H_{36}O_{9}$, was deduced from the ¹³C and ¹H NMR spectra and confirmed by HRMS. The

Table 1. ¹ H-NMR Data for Compounds 2–4, 6 (500 MHz), and 5 (400	0 MHz) ^a
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proton	compound					
	2	3	4	5	6	
H-2	3.55 (8, 2)	3.56 (12, 2)				
H-3	5.10 (8)	5.09 (12)	5.14 s	5.09 s	4.87 s	
H-5	3.23 (8, 3)	3.24 (12, 5)	3.24 (10, 5)	3.15 (9, 3.5)	3.28 (9, 4)	
H-6	2.34 m	2.34 m	2.35 m	2.31 m	2.36 m	
H-9	1.92 m	1.90 m	1.86 (13, 3.7)	1.90 m	2.27 m	
H-11	1.82 m	1.78 m	1.81 m	1.77 m	1.17 m	
	1.92 m	1.90 m	1.90 m	1.90 m	1.62 m	
H-12	1.20 m	1.18 m	1.22 m	1.20 m	1.42 m	
	1.95 m	1.96 m	1.97 m	1.95 m	1.67 m	
H-14	1.62 m	1.58 (12.5, 6)	1.61 (13, 6)	1.65 (13, 5)	2.25 m	
H-15	2.78 (16, 6)	2.80 (18, 6)	2.83 (16, 6)	2.81 (16, 5)	2.84 m	
	3.63 (16, 4)	3.64 (18, 12.5)	3.50 (16, 13)	3.49 (16, 13)	2.90 m	
H-17	5.23 s	5.18 s	5.18 s	5.25 s	5.69 s	
H-18	1.02 s	1.01 s	1.00 s	1.00 s	1.09 s	
H-19	1.07 s	1.06 s	1.17 s	1.15 s	1.23 s	
H-21	7.51 (1.5, 0.7)	7.48 m	7.49 (1.8, 0.8)	7.53 bs	7.81 (1.7)	
H-22	6.46 (1.5, 0.7)	6.45 m	6.44 (1.8, 0.8)	6.47 (2)	6.46 (1.7)	
H-23	7.43 t (2)	7.43 m	7.43 t (1.8)	7.45 t (2)	7.42 t (1.7)	
H-28	0.81 s	0.81 s	0.81 s	0.79 s	0.81 s	
H-29	0.78 s	0.78 s	0.79 s	0.75 s	0.76 s	
H-30	3.31 (2)	3.31 (2)	3.51 s	3.52 s	5.38 t (2)	
$-0CH_3$	3.73 s	3.72 s	3.74 s	3.72 s	3.72 s	
H-2′		2.75 m				
H-3′		1.28 $(7)^{b}$	7.04 qq (6.5, 1.5)			
2'-Me		1.26 $(7)^{b}$	1.97 t(1)			
3'-Me			1.93 (6.5, 1)			
-OH			4.01 s	4.15 s		
$-COCH_3$	2.24 s			2.28 s	2.13 s	

^{*a*} Chemical shifts (relative to TMS) are in ppm and coupling constants (in parentheses) in Hz. ^{*b*} Assignments interchangeable. Where signal multiplicity is not noted, d is implied. The assignments were made by a combination of COSY, HETCOR, and HMBC.

carbon			compound		
	2	3	4	5	6
C-1	214.04	214.17	213.00	213.15	215.40
C-2	48.61	48.86	78.37	77.95	76.71
C-3	77.21	76.74	84.79	84.70	84.80
C-4	39.18	39.39	40.12	39.73	38.42
C-5	42.48	42.57	42.32	42.31	41.02
C-6	33.06	33.14	32.94	32.77	32.56
C-7	174.19	174.16	173.86	174.02	174.30
C-8	60.46	60.65	63.14	62.89	136.46
C-9	55.81	55.82	55.14	55.23	56.25
C-10	48.18	48.25	49.11	48.95	48.81
C-11	19.35	19.40	19.44	19.27	19.10
C-12	33.42	33.38	33.20	33.31	33.74
C-13	36.42	36.40	36.23	36.31	36.08
C-14	45.93	45.84	45.24	45.56	44.80
C-15	33.74	33.92	33.49	33.29	29.90
C-16	172.06	171.94	171.21	171.74	169.25
C-17	78.76	78.84	78.89	78.71	77.10
C-18	26.27	26.42	26.26	26.04	21.90
C-19	16.02	15.88	16.12	16.21	15.70
C-20	120.05	120.07	120.17	120.05	120.35
C-21	141.00	140.94	140.89	141.06	141.88
C-22	110.24	110.24	110.10	110.19	109.70
C-23	143.07	143.09	143.11	143.14	142.94
C-28	20.67	20.96	20.53	20.00	19.52
C-29	22.50	22.45	22.00	21.79	21.80
C-30	63.57	63.33	67.38	67.72	129.10
$-0CH_3$	52.34	52.32	52.39	52.43	52.20
C-1′		176.02	166.92		
C-2′		34.17	127.77		
C-3′		19.45^{b}	139.75		
2'- <i>C</i> H ₃		18.94^{b}	12.60		
3'- <i>C</i> H ₃			14.63		
$COCH_3$	169.82			169.54	171.20
COCH ₃	20.80			20.80	20.32

^{*a*} Chemical shifts (relative to TMS) are in ppm and coupling constants (in parentheses) in Hz. ^{*b*} Assignments interchangeable. Where signal multiplicity is not noted, d is implied. The assignments were made by a combination of COSY, HETCOR, and HMBC.

IR spectrum indicated hydroxyl (3450 $\rm cm^{-1}),$ carbonyl (1730 $\rm cm^{-1}),$ and furan (875 $\rm cm^{-1})$ moieties.

Both the $^{13}\mathrm{C}$ and $^{1}\mathrm{H}$ spectra were closely similar to those of humilinolide D except that H-5 was now a

Table 3. Antifeedant Activity of Compounds **1**–**4** Bioassayed with *S. frugiperda* (Final Instar Larvae)⁵

compound	AI ^a SEM
methyl angolensate (1) xylocarpin (2) ruageanin A (3) ruageanin B (4)	$\begin{array}{c} 66.4 \pm 10.63 \\ 77.8 \pm 6.90 \\ 72.6 \pm 19.60 \\ 86.3 \pm 6.41 \end{array}$

^{*a*} AI represents the antifeedant index calculated from AI = [(C - T)/(C + T)]100. *C* and *T* represent the amount eaten by the larvae of the control and treatment disks, respectively (Wilcoxon's matched pairs test p < 0.05).

double doublet ($\delta_{\rm H}$ 3.28 J = 9, 4 Hz) and C-6 was present as a methylene ($\delta_{\rm H}$ 2.36, m). The C-8/C-30 trisubstituted olefin was apparent ($\delta_{\rm C}$ 136.46, 129.10; $\delta_{\rm H}$ 5.38, t, 2.0 Hz), and typically H-30 showed allylic coupling with both H-9 and H-14. The carbonyl of the –OAc group ($\delta_{\rm C}$ 171.2) and the quaternary carbinol at C-2 ($\delta_{\rm C}$ 76.71) showed HMBC with H-3 ($\delta_{\rm H}$ 4.87, s). C-3 ($\delta_{\rm C}$ 84.40) in turn showed ³ $J_{\rm CH}$ correlation with the protons of the *gem*-dimethyl protons at C-4. The stereochemistry was assigned by analogy with humilinolide D.⁸

Five of these compounds were subjected to an antifeedant bioassay^{5,6} on the final instar larvae of *Spodoptera frugiperda* at concentrations of 1000 ppm. Compounds **1**–**4** showed comparable antifeedant activity (Table 3) which was lower than that for azadiractin⁹ but more active than a model compound based on jodrellin A.⁶ Compound **5** exhibited low phagostimulation.

Experimental Section

General Experimental Procedures. IR spectra were obtained on a Pye Unicam SP3-200 spectrophotometer in Nujol mulls; optical rotations were measured with a Schmidt and Haensch Polartronic-D polarimeter in CHCl₃ solutions. ¹H and ¹³C spectra were recorded on a Varian XL 400 or a Varian Unity 500 spectrometer in CDCl₃ solutions with TMS as internal standard.

Mass spectra data were recorded on a VG 70-250S mass spectrometer operating at 70 eV (direct insertion). Silica gel 60 $PF_{254+366}$ (Merck) was used for vacuum liquid chromatography (VLC), TLC, and PTLC. Flash chromatography was done on silica gel 60 (230–400 mesh, Merck).

Plant Material. The plant material was collected in February 1992 at 2060 m in Cartago Costa Rica by Dr. T. D. Pennington and P. E. Owen of The Royal Botanic Gardens, Kew, England where a voucher specimen has been deposited (Pennington 13605).

Extraction and Isolation. The dried ground fruits (341.1 g) were exhaustively extracted with benzene to yield a dark-green viscous oil (91.9 g). The latter was washed with petroleum ether (60-80 °C) and triturated with EtOAc to yield a brown gum on evaporation of the solvent (26.6 g). The EtOAc-soluble fraction was subjected to extensive VLC, flash chromatography, and preparative TLC to yield the reported compounds. Yields are quoted as percentages based on the dry weight of ground plant material.

The EtOAc extract was subjected to VLC using $CHCl_3$ with increasing quantities of EtOAc as eluent. Fractions were grouped on the basis of their TLC profiles into three major fractions (F1–F3). F1 (90% $CHCl_3/$ 10% EtOAc) and F2 (90% $CHCl_3/10\%$ EtOAc) were subjected to flash chromatography using the same

solvent system as used for VLC. F1 yielded subfractions containing a mixture of two compounds which were further separated by PTLC (CHCl₃–EtOAc, 10:1, \times 2) into methyl angolensate (1) (0.01%) and xylocarpin (2) (0.008%). A later fraction from the flash column of F1 yielded ruageanin A (3) (0.01%) which was finally purified by PTLC (CHCl₃–EtOAc, 8:1, \times 3). F2 yielded ruageanin B (4) (0.03%) purified by PTLC (petroleum ether–EtOAc, 2:1, \times 2) and repeated recrystallization from EtOAc/MeOH. F3 (85% CHCl₃/15% EtOAc) was subjected to PTLC (CHCl₃–EtOAc, 8:1, \times 3) to yield ruageanin C (5) (0.06%) and a more polar band which on further PTLC (hexane–EtOAc, 2:1 \times 4) gave ruageanin D (6) (0.004%).

Methyl angolensate (1): crystalline solid (MeOH); mp 201–203.5 °C (lit.² mp 205 °C); $[\alpha]^{25}_{\rm D}$ –7.3° (c = 0.15) (lit.³ $[\alpha]^{25}_{\rm D}$ –4.2° (c = 2.14)); IR $\nu_{\rm max}$ 1735, 1715, 875 cm⁻¹; EIMS m/z [M⁺] 470 (100), 455 (5), 439 (4), 397 (4), 375 (25), 359 (33), 332 (26), 243 (17), 227 (9), 210 (28); HREIMS m/z 470.2292, calcd for C₂₇H₃₄O₇ 470.2305.

Xylocarpin (2): amorphous solid; $[\alpha]^{25}{}_{\rm D}$ -79.8° (c = 0.25) (lit.⁴ $[\alpha]^{25}{}_{\rm D}$ -88°); IR $\nu_{\rm max}$ 1730, 878 cm⁻¹; EIMS m/z [M⁺] 528 (14), 510 (3), 466 (3), 435 (3), 414 (2), 390 (22), 330 (8), 302 (20), 281 (25), 239 (26), 221 (100), 149 (46), 69 (67); HREIMS m/z 528.2878, calcd for C₂₉H₃₆O₉ 528.2859.

Ruageanin A (3): amorphous solid; $[\alpha]^{25}{}_{\rm D}$ -63.9° (*c* = 0.74); IR $\nu_{\rm max}$ 1730, 875 cm⁻¹; EIMS m/z [M⁺] 556 (23), 468 (7), 437 (4), 418 (28), 390 (7), 348 (8), 330 (25), 309 (52), 286 (8), 257 (8), 239 (30), 221 (100), 71 (33); HREIMS m/z [M⁺] 556.2681, calcd for C₃₁H₄₀O₉ 556.2672.

Ruageanin B (4): crystalline solid (EtOAc/MeOH); mp 227–229 °C; $[\alpha]^{25}_{D}$ –17.7 (*c* = 0.52); IR ν_{max} 3500, 1735, 1645, 880 cm⁻¹; EIMS *m*/*z* [M⁺] 584 (5), 566 (1), 502 (1), 485 (2), 467 (6), 425 (3), 407 (2), 289 (2), 224 (6), 196 (6), 164 (10), 141 (10), 121 (9), 95 (24), 83 (100), 67 (7), 55 (66); HREIMS *m*/*z* [M⁺] 584.2629, calcd for C₃₂H₄₀O₁₀ 584.2621.

Ruageanin C (5): crystalline solid (MeOH); mp 217.5–219 °C; $[\alpha]^{25}_{D}$ –15.9° (c = 0.64); IR ν_{max} 3510, 1735, 1645, 875 cm⁻¹; EIMS m/z [M⁺] 544 (49), 529 (1), 502 (2), 484 (16), 453 (2), 406 (34), 389 (3), 297 (14), 243 (26), 224 (46), 196 (47), 164 (67), 137 (86), 122 (100), 95 (84); HREIMS m/z [M⁺] 544.2311, calcd for C₂₉H₃₆O₁₀ 544.2308.

Ruageanin D (6): amorphous solid; $[\alpha]^{25}{}_{\rm D}$ -55.9° (*c* = 0.15); IR $\nu_{\rm max}$ 3450, 1730, 875 cm⁻¹; EIMS m/z [M⁺] 528 (9), 486 (6), 447 (12), 429 (11), 411 (23), 379 (19), 343 (6), 245 (11), 196 (17), 173 (14), 149 (31), 91 (47), 83 (100), 89 (63), 55 (90); HREIMS 528.2347, calcd for C₂₉H₃₆O₉ 528.2359.

Feeding Inhibition Assay on the Final Instar Larvae of *S. frugiperda.* The method used was that developed by Simmonds et al.⁵ The test insects were final instar larvae of *S. frugiperda* (J. E. Smith) (Lepidoptera:Noctuidae) which were reared in the laboratory at 25.0 \pm 1.0 °C. Larvae were fed on young leaves of maize (*Zea mays*) and their development monitored daily. The larvae were 24–36 h into the final instar and had been deprived of food for 4 h prior to being individually placed in Petri dishes.

Compounds were assayed for antifeedant activity by presentation on glass fiber disks (Whatman GF/A or GF/

C, 2.1 cm diameter) to final instar S. frugiperda larvae. The test compounds were dissolved in ethanol and assayed at 1000 ppm. Disks were made palatable by the addition of 100 μ L of sucrose (0.05 M).

Ten Petri dishes each containing one larva and two glass fiber disks [control (C) and treatment (T)] were used for every compound. The control disk (C) had 100 μ L of sucrose and 100 μ L of ethanol, while the treatment disk (T) had sucrose and a 100 μ L aliquot of one of the test solutions (compound + ethanol). Both disks were dried and weighed before being presented to the larva. The duration of the bioassay was 18 h after which the disks were reweighed and the amount of each disk consumed noted.

The antifeedant index (AI) was calculated from

$$[(C - T)/(C + T)]100$$

where C and T are the amounts of the control and treatment discs eaten, respectively. This index identified both phagostimulants (-AI values) and antifeedants (+AI values).

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